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Simulation of interdiffusion in between compartments having heterogeneously distributed donors and acceptors

E. Tüzel, K.B. Kisacikoğlu, Ö. Pekcan*

Department of Physics, Faculty of Sciences and Letters, Istanbul Technical University, Maslak 80626 Istanbul, Turkey

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Abstract

The final stage of latex film formation was simulated by introducing donors and acceptors into the adjacent compartments of a cube. Homogeneous and/or heterogeneous donor–acceptor distributions were chosen for different types of simulations. The interdiffusion of the donors and the acceptors within these cubes was generated using the Monte-Carlo technique. The decay of the donor intensity $I(t)$ by direct energy transfer (DET) was simulated for several interdiffusion steps. Gaussian noise was added to the $I(t)$ curves to obtain more realistic decay profiles. $I(t)$ decay curves were fitted to the phenomenological equation to calculate the fractional mixing at each interdiffusion step. The reliability of the Fickian diffusion model in the case of heterogeneous and homogeneous donor–acceptor distributions are discussed for latex film formation. $© 2000$ Elsevier Science Ltd. All rights reserved.

Keywords: Interdiffusion; Energy transfer; Heterogeneous distribution

1. Introduction

Polymer latex particles have been utilized in a wide variety of applications in the coating and adhesive technologies, biomedical field, information industry and microelectronics. In many of these applications, e.g. coatings and adhesives, latexes form thin polymer films on a substrate surface. Properties (mechanical, optical, transport, etc.) of the final film should be tailor-made according to the application.

Film formation from latex particles is a complicated, multistage phenomenon and depends strongly on the characteristics of colloidal particles. In general, aqueous or nonaqueous dispersions of colloidal particles, with glass transition temperature (T_g) above the drying temperature, are named hard latex dispersion, however aqueous dispersion of colloidal particles with T_g below the drying temperature is called soft latex dispersion. The term "latex film" normally refers to a film formed from soft particles where the forces accompanying the evaporation of water are sufficient to compress and deform the particles into a transparent, void-free film [1,2]. However, hard latex particles remain essentially discrete and undeformed during the drying process. Film formation from these dispersions can occur in several stages. In both cases, the first stage corre-

Transmission electron microscopy (TEM) has been used to examine the morphology of dried latex films [5,6]. These studies have shown that in some instances, the particle boundaries disappeared over time, but in other cases, the boundaries persisted for months. It was suggested that in the former case particle boundaries were healed by polymer diffusion across the junction. In the last few years, it has become possible to study latex film formation at the molecular level. Small-angle neutron scattering (SANS) was used to examine deuterated particles in a protonated matrix. It was observed that the radius of the deuterated particle increased in time as the film was annealed [7] and as the polymer molecules diffused out of the space to which they were originally confined. The process of interparticle

^{*} Corresponding author. Tel.: $+ 90-212-285-3213$; fax: $+ 90-212-285-$ 6366.

sponds to the wet initial state. Evaporation of solvent leads to second stage in which the particles form a closed pack array, here if the particles are soft they are deformed to polyhedrans (see Fig. 1). Hard latex however stays undeformed at this stage. Annealing of soft particles causes diffusion across particle–particle boundaries which leads the film to a homogeneous continuous material. In the annealing of the hard latex system, however deformation of particles first leads to void closure [3,4] and then after the voids disappear, diffusion across particle–particle boundaries starts, i.e. the mechanical properties of hard latex films can be evolved by annealing; after all the solvent has evaporated and all voids have disappeared.

Fig. 1. A pictorial representation of the stages of latex film formation from soft polymer particles. (a) The latex dispersion. (b) The solvent evaporates leaving the particles in close contact. (c) Deformation and packing of the particles. (d) Further coalescence produces a mechanically rigid film.

polymer diffusion has been studied by the direct energy transfer (DET) method, using transient fluorescence (TRF) measurements [8,9] in conjunction with latex particles labelled with donor and acceptor chromophores. The steady state fluorescence (SSF) method combined with DET was also used for studying film formation from hard latex particles [10–13]. An extensive review of the subject is given in Ref. [14]. In DET measurements distribution of donors and acceptors are thought to be crucial, i.e. it is believed that donors and acceptors have to be distributed randomly in the latex particles for the reliable TRF measurements, to determine the diffusion coefficients, *D* of polymer chains.

Recently we have performed various experiments with the photon transmission method using an UV–Visible (UVV) spectrophotometer to study latex film formation from PMMA and PS latexes where void closure and interdiffusion processes at the junction surfaces are studied [15–18]. All these studies indicate that annealing leads to polymer diffusion and mixing as the particle junction heals during latex film formation. Recently, Monte Carlo simulation of interdiffusion and its monitoring by DET during latex film formation has also been studied in our laboratories [19,20].

In this work, the Monte Carlo method was used to simulate the final stage of film formation by introducing donors and acceptors into the adjacent compartments of a cube. Four different combinations of donor–acceptor distributions were chosen for the different types of simulations. For example in the first case distribution of donors and acceptors in their adjacent compartments are taken as homogeneous and gaussian, respectively. In the second case distributions are switched from one compartment to the other. In the third case, both distributions are taken as gaussian and in the final case, distribution of donors and acceptors are both taken homogeneously to compare this case with the others.

The interdiffusion of donors and acceptors between these adjacent compartments was randomly generated by Monte Carlo method. The decay of the donor intensity, $I(t)$ by DET was simulated for several interdiffusion steps and a gaussian noise was added to generate the realistic time resolved fluorescence data. $I(t)$ decays were fitted to the phenomenological equation to obtain the fractional mixing at each interdiffusion step. The reliability of the Fickian model and the effect of heterogeneous donor–acceptor distributions are discussed at the last stage of latex film formation process.

2. DET and fluorescence decay

Polymer diffusion obeys de Gennes scaling laws for times short compared to the tube renewal time t_{tr} , but for long times it is like a random walk process (Fickian diffusion). In order to be able to determine whether the diffusion is Fickian, one must compare the experimental data with the results of simulations of DET with Fickian diffusion.

TRF in conjunction with the DET method, monitors the extent of interdiffusion of donor (D) and the acceptor (A) labelled polymer molecules. The sample is made of a mixture of D and A, labelled latex spheres. When this sample is annealed for a period of time and the donor fluorescence profiles are measured, each decay trace provides a snapshot of the extent of interdiffusion [9]. A film sample after annealing was considered to be composed of three regions; unmixed D, unmixed A and the mixed D–A region. This model was first empirically introduced by the twocomponent donor fluorescence decay [21,22].

When donor dyes are excited using a very narrow pulse of light, the excited donor returns to the ground state either by emitting a fluorescence photon or through the non-radiative mechanism. For a well-behaved system, after exposing the donors with a short pulse of light, the fluorescence intensity decays exponentially with time. However, if acceptors are present in the vicinity of the excited donor, then there is a possibility of DET from the excited donor to the ground state acceptors. In the classical problem of DET, neglecting back transfer, the probability of the decay of the donor at r_k due to the presence of an acceptor at r_i is given by [23]

$$
P_k(t) = \exp[-t/\tau_0 - w_{ik}t] \tag{1}
$$

where w_{ik} is the rate of energy transfer given by Förster as

$$
w_{ik} = \frac{3}{2} \kappa^2 \frac{1}{\tau_0} \left(\frac{R_0}{r_{ik}} \right)^6
$$
 (2)

Here R_0 represents the critical Förster distance and κ is a dimensionless parameter related to the geometry of interacting dipole. If the system contains N_D donors and N_A acceptors, then the donor fluorescence intensity decay can be derived from Eq. (2) and given by [16]

$$
\frac{I(t)}{I(0)} = \exp(-t/\tau_0) \frac{1}{N_D} \int n_D(r_k) dr_k
$$

$$
\times \prod_{i=1}^{N_A} \frac{1}{N_A} \int n_A(r_i) dr_i \exp(-w_{ik}t)
$$
(3)

Fig. 2. Several snapshots of the interdiffusion process between adjacent compartments of a cube in which donors and acceptors are distributed homogeneous and gaussian wise. (a) $K = 0.0$; (b) $K = 0.3$; and (c) $K =$ 1:0:

Here n_D and n_A represent the distribution functions of donors and acceptors. In the thermodynamic limit Eq. (3) becomes [16]

$$
\frac{I(t)}{I(0)} = \exp(-t/\tau_0) \frac{1}{N_D} \int n_D(r_k) dr_k
$$

$$
\times \exp\left(-\int n_A(r_i) dr_i (1 - \exp(-w_{ik}t))\right)
$$
(4)

This equation can be used to generate donor decay profiles by Monte Carlo techniques. It is shown that Eq. (4) reduces to a more simple form, which can be compared to the experimental data [3]. Their argument is summarized below for clarity. Changing to the coordinate $r_{ik} = r_i - r_k$ leads to

$$
\frac{I(t)}{I(0)} = \exp(-t/\tau_0) \frac{1}{N_{\rm D}} \int n_{\rm D}(r_k) dr_k
$$

$$
\times \prod_{i=1}^{N_{\rm A}} \int_{r_{\rm K}}^{R_{\rm g} - r_{\rm K}} n_{\rm A}(r_{ik} + r_k) dr_{ik} \exp(-w_{ik}t)
$$
 (5)

where R_g is an arbitrary upper limit. Placing a particular donor at the origin and assuming that the mixed and unmixed regions are created during interdiffusion of D and A, Eq. (5) becomes

$$
\frac{I(t)}{I(0)} = B_1 \exp(-t/\tau_0) \prod_{i=1}^{N_A} \frac{1}{N_A} \int_0^{R_g} n_A(r_{ik}) dr_{ik} \exp(-w_{ik}t)
$$

+ $B_2 \exp(-t/\tau_0)$ (6)

where

$$
B_{1,2} = \frac{1}{N_{\rm D}} \int_{1,2} n_{\rm D}(r_k) \, \mathrm{d}r(k) \tag{7}
$$

represent the fraction of donors in mixed and unmixed regions, respectively. The integral in Eq. (6) produces a Förster type of function [24,25]

$$
\prod_{i=1}^{N_{\rm A}} \frac{1}{N_{\rm A}} \int_0^{R_{\rm g}} n_{\rm A}(r_{ik}) \, \mathrm{d}r_{ik} \, \exp(-w_{ik}t) = \exp\left(-C\left(\frac{t}{\tau_0}\right)^{1/2}\right) \tag{8}
$$

where *C* is proportional to acceptor concentration. Eventually, one gets the following formula for the fluorescence intensity

$$
\frac{I(t)}{I(0)} = B_1 \exp\left(-t/\tau_0 - C\left(\frac{t}{\tau_0}\right)^{1/2}\right) + B_2 \exp(-t/\tau_0) \quad (9)
$$

Here it is useful to define the mixing ratio *K* representing the order of mixing during interdiffusion of the donors and the acceptors as

$$
K = \frac{B_1}{B_1 + B_2} \tag{10}
$$

Fig. 3. Several snapshots of the interdiffusion process between adjacent compartments of a cube in which donors and acceptors are distributed gaussian and homogeneous wise. (a) $K = 0.0$; (b) $K = 0.3$; and (c) $K = 1.0$.

Fig. 4. Several snapshots of the interdiffusion process between adjacent compartments of a cube in which both donors and acceptors are distributed gaussian wise. (a) $K = 0.0$; (b) $K = 0.3$; and (c) $K = 1.0$.

Donors Acceptors

Fig. 5. Several snapshots of the interdiffusion process between adjacent compartments of a cube in which both donors and acceptors are distributed homogeneously. (a) $K = 0.1$; (b) $K = 0.4$; and (c) $K = 1.0$.

3. Simulation of interdiffusion

The interdiffusion of donors and acceptors between two adjacent compartments corresponds to the last stage of latex film formation process. Here the geometry is simplified using cubes instead of the polyhedrons, and donors and acceptors are randomly distributed in separate adjacent compartments in a cube. Figs. 2a, 3a, 4a and 5a present the four types of combinations of donor–acceptor distributions. In Fig. 2a donors and acceptors are distributed in the adjacent compartments in homogeneous and gaussian wise distributions, respectively. When these distributions are completely inversed, the situation is presented in Fig. 3a. Figs. 4a and 5a present acceptors and donors both distributed in separate compartments in gaussian and in homogeneous wise distributions, respectively.

Figs. 2b, 3b, 4b and 5b present the picture after the Brownian motion of donors and acceptors generated for several interdiffusion steps for each combination of donor–acceptor pairs which are given in Figs. 2a, 3a, 4a and 5a, respectively. In each diffusion step, all the donors and acceptors move within a range of $0-1$ Å in any direction, but are reflected from the sides of the cube. After each diffusion step, the diffusion time increments one unit. 25×10^3 diffusion steps were used for all sample simulations. The decay of donor intensity by DET is simulated for the configurations at the end of each 100 step of diffusion, therefore the diffusion process can be monitored quite clearly and accurately. Moreover, the average is taken over 10 different runs for each initial distribution. Figs. 2c, 3c, 4c and 5c present the final picture of the interdiffusion between two adjacent compartments in a cube.

The donor decay profiles were generated using Eq. (4). The side of the cube, L , is taken as 500 \AA and the Förster distance as 26 \AA . The number of donors, N_D , and acceptors N_A , are both chosen as 500. The w_{ik} values for each donor– acceptor pair are obtained from Eq. (2). The parameter κ^2 is chosen as 0.476, a value appropriate for immobile dyes [20], and the donor lifetime τ_d is taken as 44 ns. Eq. (4) is then used to simulate the donor intensity $I(t)$ decay profiles. $I(0) = 2 \times 10^4$ is chosen and the decay profiles are obtained for a 250 ns interval, divided into 250 channels of 1 ns each. Decay profiles at the several interdiffusion steps for both donors and acceptors are homogeneously distributed in adjacent compartments are presented in Fig. 6a.

Here, one may also take into account the effect of the lamp profile when calculating the decay profiles [19,20]. To do so the decay profiles generated by the Monte Carlo simulation should be convolved with an experimental lamp profile, then the experimentally measured $\phi(t)$ is obtained by convolution of $I(t)$ with the instrument response function $L(t)$, as

$$
\phi(t) = \int_0^t L(t)I(t-s) \, \mathrm{d}s \tag{11}
$$

In this work, since we are interested in the effect of donor– acceptor distributions on the interdiffusion, instead of using

Fig. 6. (a) Decay curves at the several interdiffusion steps for both donors and acceptors are homogeneous distributed in adjacent compartments. (I) $K = 0.1$; (II) $K = 0.5$; (III) $K = 1.0$. (b) Noisy decay curves for the above picture.

experimental decay profiles we used generated decay profiles. This assumption is valid if one uses a delta, δ function light source (e.g. a very fast laser) as the lamp profile. In this case no convolution is needed and Eq. (11) produces $I(t)$. However, to obtain more realistic decay profiles, gaussian noise can be added to the original decay profiles using the Box, Muller and Marsaglia [24] algorithm. In this algorithm, at first two gaussian numbers (U_1, U_2) between 0 and 1 are created. Then V_1 and V_2 are calculated as shown below

$$
V_1 = 2U_1 - 1 \tag{12}
$$

Fig. 7. Comparison of the plots of the mixing ratio *K* versus diffusion time for different initial distributions as: (a) donors and acceptors are distributed homogeneous and gaussian wise; (b) donors and acceptors are distributed gaussian and homogeneous wise; (c) both donors and acceptors are distributed gaussian wise; (d) both donors and acceptors are distributed homogeneously.

$$
V_2 = 2U_2 - 1\tag{13}
$$

Both V_1 and V_2 are distributed randomly in the range $[-1,1]$. *S* is calculated from these two numbers

$$
S = V_1^2 + V_2^2 \tag{14}
$$

If $S < 1$, operation is unsuccessful and new U_1 and U_2 numbers are created. If $S > 1$, X_1 and X_2 are calculated as shown below

$$
M = q \left(\frac{-2 \ln S}{S}\right)^{1/2} \tag{15}
$$

$$
X_1 = (V_1 M) + p \tag{16}
$$

$$
X_2 = (V_2M) + p \t\t(17)
$$

 X_1 and X_2 are mutually independent. They are gaussian numbers with an average *p* and standard deviation *q*. The noisy decay profiles for the homogeneously distributed donors and acceptors at several interdiffusion steps are shown in Fig. 6b.

In order to calculate the mixing ratio, *K* defined in Eq. (10) one should fit the generated decay profiles to Eq. (9). The decay profiles were fitted to Eq. (9) using the Levenberg–Marquart [25] algorithm. During fits, the parameters *C* and τ_0 are kept constant $(C = 1)$ and only the parameters B_1 and B_2 are varied. More than 5000 decay profiles are fitted and the goodness of fitting is varied around χ^2 < 1.5. The produced B_1 and B_2 values are used to obtain *K* values at each interdiffusion step. Fig. 7 compares the plots

Fig. 8. The plots of $\ln(1 - K)$ versus diffusion time obtained for different combinations of distributions given in Fig. 7. The solid lines present the fit of the data to Eq. (19). The slope of the solid lines produced diffusion constants, which are listed in Table 1. The regions used for the linear fits are shown within arrows in Figs. 8a–d. In Fig. 8b all the data points are used in the fit.

of *K* versus diffusion time for the interdiffusions presented in Figs. 2–5. Each curve in Fig. 7 is obtained from the average of a set of 10 runs.

To test whether the simulated interdiffusion is Fickian or

Table 1

 $D\pi^2/a^2$ values are produced by fitting the data in Fig. 7 to Eq. (19). The fits are presented in Fig. 8 for the various combinations of distributions. R^2 is the correlation coefficient for the fits

Donor	Acceptor	$D\pi^2/a^2$ ($\times 10^{-4}$)	R^2
Homogeneous Gaussian Gaussian Gaussian Homogeneous	Homogeneous Gaussian Homogeneous	1.14 ± 0.01 1.37 ± 0.03 1.41 ± 0.03 , 0.66 ± 0.02 0.91 ± 0.01	0.995 0.925 0.973, 0.958 0.991

not, the planar sheet model is chosen [26]. In this model the fraction of the diffusing substance that has diffused out of the planar sheet at time *t* is given by

$$
K_{\rm s} = \frac{8}{\pi} \sum_{n=0} \frac{1}{(2n+1)^2} \exp\left(-\frac{D(2n+1)^2 \pi^2 t}{a^2}\right) \tag{18}
$$

where D is the diffusion constant and a is the maximum distance over which diffusion can occur. Since $\lim_{k\to\infty} K_s =$ 1, Eq. (18) can be written for $n = 0$ in the form

$$
\ln(1 - K_s) = -\frac{D\pi^2 t}{a^2}
$$
 (19)

 $ln(1 - K)$ values are plotted versus diffusion time in Fig. 8 and were fitted to Eq. (19). The fits obtained for all of the

four combinations of distributions are shown in Fig. 8a–d. The solid lines in the plots represent the fitting curve and the dots represent the digitized data. The diffusion constants and the correlation coefficients showing the goodness of fits are presented in Table 1.

4. Conclusions

Fits in Fig. 8 and the values in Table 1 strongly suggest that people who work in the TRF area have to be very careful to synthesize their latex particles which are labelled with the fluorescence dyes. In this work, it is observed that when the dye distribution is not homogeneous, different results in interdiffusion processes can be produced and even the latex particles are in equal size. All data in Fig. 8 present that interdiffusion saturates at the long time region. At the short time region the initial donor–acceptor distribution is quite critical and effects the interdiffusion (mixing ratio, *K*). When donors are distributed gaussian wise, delay for the onset of interdiffusion is observed at the early time region, which is obvious, since it takes some time for the donors to reach the acceptors to perform DET. In this case if the acceptors are distributed homogeneously, interdiffusion occurs with a single diffusion constant *D*, however if the acceptors are distributed gaussian wise, two different interdiffusion regimes can be observed at the intermediate time region. In other words, after a certain delay at early times, donors and acceptors meet quite fast to perform DET and then interdiffusion slows down and finally mixing saturates at longer times.

When the donors are distributed homogeneously the delay at the short time region is quite small, especially if the acceptors are distributed gaussian wise, no delay is observed. In this case when the acceptors are distributed either gaussian or homogeneous wise, a single interdiffusion regime is observed at the intermediate time region where in both cases the interdiffusion rate is similar and much smaller than when the donors are distributed gaussian wise (see Table 1).

In conclusion, if one assumes that the ideal distribution for donors and acceptors in latex particles are both homogeneous, then one has to expect that experimental results for *K* should obey the picture in Fig. 8d, even though the picture in Fig. 8a looks much better, i.e. interdiffusion starts with no delay and produces single interdiffusion constants.

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References

- [1] Eckersley ST, Rudin A. J Coatings Technol 1990;62(780):89.
- [2] Joanicot M, Wong K, Maquet J, Chevalier Y, Pichot C, Graillat C, Linder P, Rios L, Cabane B. Prog Colloid Polym Sci 1990;81:175.
- [3] Sperry PR, Snyder BS, O'Downd ML, Lesko PM. Langmuir 1994;10:2619.
- [4] Mackenzie JK, Shuttleworth R. Proc Phys Soc 1946;62:838.
- [5] Vanderhoff JW. Br Polym J 1970;2:161.
- [6] Distler D, Kanig G. Colloid Polym Sci 1978;256:1052.
- [7] Kahn K, Ley G, Schuller H, Oberthur R. Colloid Polym Sci 1988;66:631.
- [8] Winnik MA, Wang Y, Haley F. J Coatings Technol 1992;64:51.
- [9] Pekcan \ddot{O} , Winnik MA, Croucher MD. Macromolecules 1990;23:2673.
- [10] Canpolat M, Pekcan Ö. J Polym Sci: Polym Phys Ed 1996;34:691.
- [11] Canpolat M, Pekcan Ö. Polymer 1995;36:4433.
- [12] Pekcan Ö, Canpolat M. J Appl Polym Sci 1996;59:277.
- [13] Canpolat M, Pekcan Ö. Polymer 1995;36:2025.
- [14] Pekcan Ö. Trends Polym Sci 1994;2:236.
- [15] Pekcan Ö, Kemeroğlu F, Arda E. Eur Polym J 1998;34:1371.
- [16] Pekcan Ö, Kemeroğlu F. J Appl Polym Sci 1999;72:981.
- [17] Pekcan Ö, Arda E, Kesenci K, Piskin E. J Appl Polym Sci 1998;68:1257.
- [18] Pekcan Ö, Arda E. J Appl Polym Sci 1998;70:339.
- [19] Güntürk KS, Giz AT, Pekcan Ö. Eur Polym J 1998;34:789.
- [20] Güntürk KS, Giz AT, Pekcan Ö. Polymer 1998;39:10.
- [21] Winnik MA, Pekcan Ö, Croucher MD. In: Condon F, Otterwell RH, editors. Scientific methods for the study of polymer colloids and their applications, NATO ASI, Dordrecht: Kluver Academic, 1988.
- [22] Wang Y, Winnik MA. Macromolecules 1993;26:3147.
- [23] Förster TH. Ann Phys 1948;2:55.
- [24] William H, Tenkolsky A, Vetterling T, Flannery P, editors. Numerical Recipes Inc. 2nd ed. Cambridge University Press, 1992.
- [25] Klafter J, Blumen A. J Chem Phys 1984;80:875.
- [26] Bauman J, Fayer MD. J Chem Phys 1986;85:408.